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S. No. of Question Paper : **6456**

Unique Paper Code : **2494001001**

Name of the Paper : **Molecules of Life (GE-1)**

Name of the Course : **B.Sc. (Hons.) Biochemistry**

Semester : **II**

Duration : **2 Hours**

Maximum Marks : **60**

(Write your Roll No. on the top immediately on receipt of this question paper.)

There are six questions.

Attempt any *four* questions.

All questions carry equal marks.

Question No. **1** is compulsory.

1. (a) Define the following terms with suitable examples wherever applicable :
- (i) Zwitterion
 - (ii) Isoelectric point
 - (iii) Chargaff's Rule
 - (iv) Eicosanoids
 - (v) Mutarotation.

P.T.O.

- (b) Match the biomolecules from List A with List B :

List A : Glyceraldehyde, Palmitic acid, Phospholipid, Linoleic acid, Reducing sugar, Adenine, Cholesterol, Cerebroside, Ribose, Glycogen

List B : Pentose, Homopolysaccharide, Fructose, Triose, PUFA, Steroid, Glycolipid, Fatty acid, Nitrogenous base, Phosphatidylethanolamine. 10,5

2. Differentiate between the following :

(a) Essential and non-essential amino acids

(b) Reducing and non-reducing disaccharide

(c) DNA and RNA

(d) Cellulose and Glycogen

(e) Glycerolipids and Sphingolipids. 3×5

3. (a) Draw the structure of the following :

(i) α -D glucopyranose

(ii) Alanine

(iii) Phosphatidic acid

(iv) Phenylalanine

(v) 2'-deoxyribose.

- (b) Explain the salient features of the structure of B-DNA using a suitable diagram.
- (c) What are non-standard amino acids ? Discuss with suitable examples (any *two*). Draw the structures and mention their significance. 5×3
4. (a) Name the amino acid that absorb ultraviolet light. Which of these absorbs maximum UV light ?
- (b) Explain the effect of unsaturation on the melting point of fatty acids using suitable examples.
- (c) Indicate whether the following pairs of sugars are epimers, anomers or enantiomers :
- (i) α -D Glucopyranose and β -D Glucopyranose
- (ii) D-Glucose and D-Mannose, and
- (iii) D-Fructose and L-Fructose.
- (d) Write an example each for the following :
- Essential fatty acid, an imino acid, a purine base, a sphingolipid, an amino acid that is optically inactive, a tetrameric protein.
- (e) Write briefly on the organization of DNA into Chromatin. 3×5
5. Mention the reasons for the following statements :
- (a) Amino acids act as buffers.

- (b) Lactose exists in two anomeric forms, while no anomeric form of sucrose has been reported.
 - (c) DNA is more stable than RNA.
 - (d) DNA rich in GC base pairs melts at a higher temperature than the DNA that is rich in AT base pairs.
 - (e) Protein which is abundant in basic amino acids exhibits a positive charge. 3×5
6. (a) Write a note explaining the levels of protein structure.
- (b) Explain the functions of RNA briefly.
- (c) Why are triglycerides considered as preferred choice as storage fuel compared to Glycogen ?
- (d) What are Coenzymes ? Name any *three* Vitamins and their corresponding coenzymes. 5,4,3,3