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Your Roll No.....

Sr. No. of Question Paper : 7032

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Unique Paper Code : 2222514701

Name of the Paper : Statistical Mechanics

Name of the Course : B.Sc. Prog. – (Physics)_NEP: UGCF-2022

Semester : VII

Duration : 3 Hours

Maximum Marks : 90

Instructions for Candidates

1. Write your Roll No. on the top immediately on receipt of this question paper.
2. Question 1 is compulsory.
3. Attempt any 4 questions out of the remaining 5 questions.
4. Symbols have their usual meaning.

Q. 1 Answer any *six* from the following:

(6 x 3 =18)

- (i) Define the term, microstate and macrostate giving one example each?
- (ii) Draw the graph for the variation of entropy (S) and specific heat (C_v) with respect to temperature ($T > 0$) for a two level system.
- (iii) For a classical system of N -atoms, energy of each atom of mass m is given by

$$\varepsilon = \beta(x^3 + y^3) + \frac{1}{2m}(p_x^2 + p_y^2) + \frac{1}{2}m\omega^2(x^2 + y^2) + \alpha(x + y)$$

where α, β and ω are constants. Determine the total energy and average energy per particle of the system using equipartition theorem.

P.T.O.

- (iv) Write three properties of liquid Helium.
- (v) Calculate the total number of possible ways of distributing 2 bosons in 3 quantum states.
- (vi) Write three properties of the particles that follows Fermi-Dirac statistics.
- (vii) Calculate the Fermi energy for silver using the following data:

Atomic weight = 108, Density = 10.5 g/cm³.

Q. 2 (a) Show that the entropy (S) and thermodynamic probability (Ω) are related as

$$S = k_B \ln \Omega \quad 6$$

- (b) A system of 4 ideal classical particles can be in any one of 4-states of energies ϵ , 2ϵ , 3ϵ and 4ϵ respectively. If total energy is 15ϵ for this system, calculate its entropy. 6

- (c) Prove that the partition function for an ideal monoatomic gas consisting of N distinguishable particles in a cubical volume V at temperature T is given by :

$$Z = V^N (2\pi m k_B T / h^2)^{3N/2} \quad 6$$

Q. 3 Consider a two level system of N distinguishable particles. Each particle can occupy only one of two energy levels of energy ϵ_1 and ϵ_2 (where $\epsilon_1 < \epsilon_2$). Particles are distributed in such a way that n_1 particles are present in level of energy ϵ_1 and n_2 particles resides in energy level ϵ_2 . (Assume N to be very large and $N = n_1 + n_2$)

- (a) Find the entropy and energy of this system. Show that entropy of this system is maximum when $n_2 = N/2$. 9
- (b) Obtain the general expression of temperature for the above mentioned isolated system and explain how it is possible to attain negative temperature for this system. 9

Q. 4. (a) Obtain Planck's radiation formula using Bose-Einstein statistics. 9

- (b) For a photon gas, show that the specific heat at constant volume (C_V) is 3 times its entropy. $C_V = 3S$ 9

Q. 5 (a) Obtain the Bose-Einstein distribution function. Show that at high temperatures, this function reduces to Maxwell-Boltzmann distribution function. 9

(b) Show that below the transition temperature ($T < T_c$), pressure of an ideal quantum gas consisting of N non-relativistic spinless Boson particles in 3 dimensions is given by

$$P(T) = 0.51(N_{ex}k_B/V)T$$

where N_{ex} is the number of bosons in the excited state and V is the volume. 9

Q. 6 (a) Using the following data of a white dwarf star, find the nature of the electron gas inside the same: a sphere consisting of helium gas of mass $M = 10^{30}$ kg with density $\rho = 10^{10}$ kg/m³ and temperature T of the order of 10^6 K. (use mass of proton $\approx 10^{-27}$ kg and mass of electron $\approx 10^{-30}$ kg) 9

(b) Calculate the temperature (in terms of Fermi temperature T_F) at which the chemical potential μ of an electron gas become zero. 9

Use:

$$\int_0^\infty \frac{x^{1/2}}{e^x + 1} = 0.678$$

Some useful constants and integrals :

$$c = 3 \times 10^8 \text{ ms}^{-1},$$

$$m_e = 9.11 \times 10^{-31} \text{ Kg},$$

$$b = 2898 \text{ } \mu\text{m K},$$

$$\sigma = 5.67 \times 10^{-8} \text{ Jm}^{-2}\text{s}^{-1}\text{K}^{-4},$$

$$h = 6.626 \times 10^{-34} \text{ J s} = 4.136 \times 10^{-15} \text{ eV s},$$

$$k_B = 1.38 \times 10^{-23} \text{ J/K} = 8.617 \times 10^{-5} \text{ eV/K},$$

$$\int_0^\infty \frac{x^{n-1}dx}{(z^{-1}e^x)-1} = \Gamma(n)g_n(z); \quad g_n(1) = \zeta(n)$$

$$\int_0^\infty \frac{x^{n-1} dx}{(z^{-1}e^x)+1} = \Gamma(n)f_n(z); f_n(z) = g_n(z) - [2^{1-n}g_n(z^2)]$$

$$\zeta(3/2) = 2.612, \zeta(5/2) = 1.341, f_{3/2}(1) = 0.765,$$

$$\zeta(2) = \frac{\pi^2}{6}, \zeta(4) = \frac{\pi^4}{90}, \Gamma\left(\frac{1}{2}\right) = \sqrt{\pi}$$